

# United States Patent and Trademark Office

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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/743,146	12/23/2003	Hitoshi Matsuoka	1691-0195P	7474	
2292 7590 10/05/2007 BIRCH STEWART KOLASCH & BIRCH PO BOX 747			EXAMINER		
			DOVE, TRACY MAE		
FALLS CHURCH, VA 22040-0747			ART UNIT	PAPER NUMBER	
		•	1745		
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			NOTIFICATION DATE	DELIVERY MODE	
			10/05/2007	ELECTRONIC	

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

mailroom@bskb.com

Office Action Summary    Application No.   Application No.   Application No.   Application No.   Art Unit   Tracy Dove   1745							
Examiner		Application No.	Applicant(s)				
Tracy Dove 1745  - The MAILING DATE of this communication appears on the cover sheet with the correspondence address Period for Reply  A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.  WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.  If NO period for regly is specified above, the maintain stability prints will apply and will expire SIX (8) MONTHS from the maining date of this communication. 1800 period for regly is specified above, the maintain stability prints will apply and will expire SIX (8) MONTHS from the maining date of this communication. 1800 period for regly is specified above, the maintain stability prints will apply and will expire SIX (8) MONTHS from the maining date of this communication. 1800 period for regly is specified above, the maintain stability prints will be regarded above. 1800 period will apply and will expire SIX (8) MONTHS from the maining date of this communication. 1800 period will apply and will expire SIX (8) MONTHS from the maining date of this communication. 1800 period will apply and will expire SIX (8) MONTHS from the maining date of this communication. 1800 period will be printed will expire size of the printed will be printed will expire size of the maining date of this communication. 1800 period will be printed will be pr	Office Action Summany		MATSUOKA ET AL.				
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WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.  Editations of them may be audited under the provisions of 37 CFR 1.13(e). In ceventh, however, may a reply be tailey filed of the SIX (6) MONTH'S from the mailing date of this communication.  HAD prend for reply in specified above, the mainimus estudiny period will apply and will expire SIX (8) MONTH'S from the mailing date of this communication.  HAD prend for reply in specified above, the mainimus estudiny period will apply and will expire SIX (8) MONTH'S from the mailing date of this communication.  Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patient them adjustment. See 37 CFR 1.704(b).  Status  1 ∑ Responsive to communication(s) filed on 12 September 2007.  2a ∑ This action is FINAL.  2b ∑ This action is non-final.  3) ∑ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213.  Disposition of Claims  4) ∑ Claim(s) 1 and 3-20 is/are pending in the application.  4a) Of the above claim(s) 3 and 4 is/are withdrawn from consideration.  5) ∑ Claim(s) is/are allowed.  6 ∑ Claim(s) is/are allowed.  6 ∑ Claim(s) is/are allowed.  6 ∑ Claim(s) is/are objected to.  8 □ Claim(s) is/are objected to restriction and/or election requirement.  Application Papers  9 ∑ The specification is objected to by the Examiner.  Application Papers  9 ∑ The provision of the control of the drawing(s) be held in abeyance. See 37 CFR 1.121(d).  11) ∑ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).  a) ∑ All b⟩ ∑ Some * ○□ None of:  1.∑ Certified copies of the priority documents have been received.  2.∑ Certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).  **Attachment(s)  10 Notice of Reference		ears on the cover sheet with the c	orrespondence address				
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3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).  * See the attached detailed Office action for a list of the certified copies not received.  Attachment(s)    Outline   Ou	1. Certified copies of the priority documents have been received.						
application from the International Bureau (PCT Rule 17.2(a)).  * See the attached detailed Office action for a list of the certified copies not received.  Attachment(s)    Outline   Outl	2. Certified copies of the priority documents have been received in Application No						
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Attachment(s)  1) Notice of References Cited (PTO-892)  Notice of Draftsperson's Patent Drawing Review (PTO-948)  Information Disclosure Statement(s) (PTO/SB/08)  Attachment(s)  Interview Summary (PTO-413) Paper No(s)/Mail Date  Notice of Informal Patent Application	application from the International Bureau (PCT Rule 17.2(a)).						
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Paper No(s)/Mail Date 6) Other:	3) Information Disclosure Statement(s) (PTO/SB/08)						
	Paper No(s)/Mail Date	6)	·				

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# **DETAILED ACTION**

This Office Action is in response to the communication filed on 9/12/07. Applicant's arguments have been considered, but are not persuasive. Claims 1 and 3-20 are pending. Claims 3 and 4 are withdrawn.

## Continued Examination Under 37 CFR 1.114

A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 9/12/07 has been entered.

#### Election/Restrictions

Applicant requested nonelected Group II be rejoined with Group I. Group II will be considered for rejoinder only if all claims in elected Group I are directed toward allowable subject matter.

## Specification

The amendment filed 8/17/07 is objected to under 35 U.S.C. 132(a) because it introduces new matter into the disclosure. 35 U.S.C. 132(a) states that no amendment shall introduce new matter into the disclosure of the invention. The added material which is not supported by the original disclosure is as follows: "the aspect ratio of the lamellar particle is the ratio of the longest diameter (the length of a portion having the longest diameter of the particle) to the thickness of the lamellar particle".

Applicant is required to cancel the new matter in the reply to this Office Action.

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# Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1, 5-14 and 17-20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Fenton et al., US 6,465,136.

Fenton teaches a composite membrane structure comprising a composite membrane and at least one protective layer disposed adjacent to the composite membrane. The composite membrane is a porous polymeric matrix (porous film) and an ionically conductive solid, noble metal or combination thereof dispersed within the matrix, and preferably, a binder. The binder is an ion exchange polymer. The protective layer comprises binder and ionically conductive solid, hygroscopic fine powder or a combination thereof (abstract). The porous polymeric matrix possesses high porosity and extremely fine pore size. Preferably the matrix has pores possessing a maximum dimension in the range from about 0.025 µm to about 1 µm. (4:66-5:13). The ionically conductive solid and binder are impregnated into the porous polymeric matrix in order to render the interior volume of the membrane occlusive (33-39). The binder is preferably present in the composite membrane and is any chemically and electrochemically stable ion exchange resin or other polymer with high ionic conductivity (6:19-29). The protective layer comprises a binder and hygroscopic fine powder. The fine powders may be silica or titania and have an average particle size less than about 10 µm (6:61-67). The binder employed in the protective layer may be the same as that employed in the composite membrane (6:40-60). The

matrix has porosity in the range of 40-95%, more preferably 60-90% (5:14-20). The matrix has a thickness of 6-102 μm (4:55-65). The binder and ionically conductive solid employed in the protective layer are the same types of materials as those dispersed within the composite membrane (continuous phase) (6:50-54). Noble metals (inorganic filler) can be used in addition to the ionically conductive solid in the composite membrane (6:7-18). Crosslinkable ion exchange materials are described at column 6, lines 25-29.

Fenton does not explicitly teach fine powders having an aspect ratio of 50 to 2000.

However, the invention as a whole would have been obvious to one having ordinary skill in the art at the time the invention was made because the courts have held that where the only difference between the prior art and the claimed invention was a recitation of relative dimensions (particle size or aspect ratio) of the claimed device (composite membrane) and a device having the claimed relative dimensions would not perform differently than the prior art device, the claimed device was not patentably distinct from the prior art device. See MPEP 2144.04.

Regarding claims 10 and 11, one of skill would have known that the composite membrane with protective layer of Fenton and the ion exchange membrane of the claimed invention would have had similar properties.

Claims 15 and 16 are rejected under 35 U.S.C. 103(a) as being unpatentable over Fenton et al., US 6,465,136 in view of Roark et al., US 7,001,446.

See discussion of Fenton above regarding claim 1. Fenton does not explicitly teach the fine particles are selected from the materials of pending claim 15.

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However, Roark teaches a protective layer is applied to a membrane to protect the catalyst from the detrimental effects of feedstream and other contaminants that may enter the membrane. Exemplary protective layers include alumina, zirconia and other metal oxides. A protective layer of porous perovskites can be used to protect the membrane from poisoning. In addition, a protective layer comprising a ceramic or other material that absorbs water or hydrocarbons can be provided (9:57-10:16).

Therefore, the invention as a whole would have been obvious to one having ordinary skill in the art at the time the invention was made because one of skill would have been motivated to use the perovskite protective layer material of Roark for the protective layer of Fenton because Fenton teaches useful powders for the protective layer are those which are capable of absorbing water. The protective layer of Fenton decreased the amount of fuel crossover and the protective layer of Roark prevents the feedstream and contaminants from passing through to the membrane. Therefore, of one of skill would have been motivated to use the protective layer materials of Roark for the protective layer of Fenton to decrease fuel (feedstream) crossover.

### Response to Arguments

Applicant's arguments filed 8/17/07 have been fully considered but they are not persuasive.

Applicant argues Fenton fails to teach or suggest an ion exchange membrane comprising a crosslinked ion exchange polymer. However, Fenton teaches crosslinkable ion exchange materials at column 6, lines 25-29. Perfluorosulfonic acid (Nafion), is a crosslinkable monomer.

Applicant argues Fenton does not teach or suggest a lamellar particle having an aspect ratio of 50-2000. However, the invention would have been obvious to one of skill in the art

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because the courts have held that where the only difference between the prior art and the claimed invention was a recitation of relative dimensions (particle size or aspect ratio) of the claimed device (composite membrane) and a device having the claimed relative dimensions would not perform differently than the prior art device, the claimed device was not patentably distinct from the prior art device. See MPEP 2144.04. Applicant asserts unexpected results are shown in the Examples of the specification. However, the Examples are not representative of the Fenton reference. Unexpected results must distinguish the claimed invention over *the prior art of record*. Furthermore, the Examples are not commensurate in scope with the claimed invention because at least claim 1 does not require a fuel cell wherein methanol is supplied as an anode reactant gas.

#### Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Tracy Dove whose telephone number is 571-272-1285. The examiner can normally be reached on Monday-Thursday (9:00-7:30).

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Pat Ryan can be reached on 571-272-1292. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent

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system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

September 28, 2007